

TEST NO. 5**CH NO. 1 TO 4 (1ST HALF BOOK)****CHEMISTRY**(Class 9th)-2020

Paper : (Objective type)

Time Allowed:- 15 Minutes

Maximum Marks:- 12

Note:- Write answers to the questions on the objective answer sheet provided. Four possible answers A,B,C and D to each question are given. Which answer you consider correct, fill the circle in front of A,B,C or D with Marker or Pen ink to each question on the answer sheet provided.

- 1.1 Mass of electron is
(a) 1.672×10^{-24} kg (b) 1.672×10^{-21} g (c) 9.106×10^{-28} kg (d) 9.106×10^{-28} g
2. Empirical formula of glucose is.
(a) CH₂O (b) CHO (c) C₂HO (d) C₂H₂O
3. How much mass is in one mole of water?
(a) 2 (b) 3 (c) 16 (d) 18
4. Who introduced the concept of orbit?
(a) J.J.Thomson (b) Rutherford (c) Bohr (d) Planks
5. The maximum electrons in M-Shell are.
(a) 2 (b) 8 (c) 18 (d) 32
6. Mendeleev's periodic table was based on.
(a) Electronic configuration (b) Atomic size
(c) Atomic number (d) Mass number
7. The distance between the nuclei of two carbon atom is.
(a) 154 Pm (b) 140 Pm (c) 110 Pm (d) 115 Pm
8. Electron negativity of carbon is.
(a) 2 (b) 1 (c) 26 (d) 4
9. After gaining one electron chlorine atom attains the electronic configuration of which noble gas?
(a) Helium (b) Neon (c) Argon (d) Krypton
10. Which bond present in C₂H₄ molecule?
(a) Single covalent bond (b) Double covalent bond (c) Triple covalent bond
(d) None of these
11. Which type of force is present in hydrogen bonding?
(a) Intermolecular force (b) Ionic force (c) Covalent force (d) Metallic force
12. Covalent bond involves the.
(a) Donation of electron (b) Acceptance electrons
(c) Sharing of electrons (d) Repulsion of electrons

	A	B	C	D		A	B	C	D		A	B	C	D		A	B	C	D	
1	(A)	(B)	(C)	(D)	4	(A)	(B)	(C)	(D)	7	(A)	(B)	(C)	(D)	10	(A)	(B)	(C)	(D)	13
2	(A)	(B)	(C)	(D)	5	(A)	(B)	(C)	(D)	8	(A)	(B)	(C)	(D)	11	(A)	(B)	(C)	(D)	14
3	(A)	(B)	(C)	(D)	6	(A)	(B)	(C)	(D)	9	(A)	(B)	(C)	(D)	12	(A)	(B)	(C)	(D)	15

نوٹ: ہر پوچھی سوال تانے کو توجہ سے پڑھیں اور ہر MCQ کی درست آپشن A,B,C,D کو پین کی سیاہی یا مارکر سے اس طرح پُر کریں کہ سیاہی دائرے سے باہر نہ نکلے۔ ایک سے

زیادہ دائروں کو پُر کرنے یا کاٹ کر پُر کرنے کی صورت میں مذکورہ جواب غلط تصور ہوگا۔